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## Structure Reports

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## Key indicators

Single-crystal X-ray study
$T=298 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.005 \AA$
$R$ factor $=0.051$
$w R$ factor $=0.147$
Data-to-parameter ratio $=13.4$
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

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# $N$-Ethyl-2-(4-nitrophenoxy)- $N$-phenylacetamide 

The asymmetric unit of the title compound, $\mathrm{C}_{16} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{O}_{4}$, obtained from the reaction of 4-nitrophenol with N -ethyl- N phenylchloracetamide in dimethylformamide at 373 K , contains two crystallographically independent molecules which differ in the orientations of the ethyl groups. The dihedral angles between the two benzene rings are 57.72 (9) and $67.78(9)^{\circ}$ in the two molecules.

## Comment

Much attention has been paid to amide-type compounds and their metal complexes for their their use in molecular recognition, ion electrodes and photochemistry, their topological structure in biochemistry, catalysis, magnetics and so on (Linton \& Hamilton, 1997; Lippard \& Berg, 1994; Saravanakumar et al., 2005). The amide linkage is central to protein structure and function in biological systems (Chang et al., 1991; Rebek, 1990). The amide linkage has been utilized to create various molecular devices for a variety of purposes in organic chemistry.

(I)

The title compound, (I), contains two independent molecules in the asymmetric unit (Fig. 1) which differ in the orientations of the ethyl groups $[\mathrm{C} 1-\mathrm{N} 1-\mathrm{C} 15-\mathrm{C} 16=$ $95.0(4)^{\circ}$ and $\left.\mathrm{C} 17-\mathrm{N} 3-\mathrm{C} 31-\mathrm{C} 32=-107.8(3)^{\circ}\right]$. The dihedral angle between the C3-C8 and C9-C14 benzene rings is 57.72 (9) ${ }^{\circ}$ and that between the C19-C24 and the C25-C30 benzene rings is $67.78(9)^{\circ}$. The corresponding bond lengths and angles in the two independent molecules agree with each other.

The crystal packing is stabilized by weak $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ interactions (Table 1).

## Experimental

Compound (I) was prepared according to the literature method of Zhang et al. (2004). Anhydrous $\mathrm{K}_{2} \mathrm{CO}_{3}(4.0 \mathrm{~g})$ was added slowly to a DMF solution ( 20 ml ) of 4-nitrophenol $(0.82 \mathrm{~g})$ at 383 K . An hour later, a DMF solution ( 20 ml ) containing $N$-ethyl- $N$-phenyl-2chloroacetylamide ( 1.18 g ) was added slowly to the mixture. The reaction mixture was stirred for 24 h at 383 K . Distilled water ( 50 ml ) was poured into the mixture, which was extracted by $\mathrm{CHCl}_{3}$. The

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## organic papers

combined organic phase was evaporated under vacuum. The crude product was chromatographed on silica gel ( EtOAc /petroleum ether, $100 \mathrm{ml}, v / v 4: 1$ ) to afford a yellow product. Single crystals of (I) were obtained by recrystallization from methanol at room temperature.

## Crystal data

$\mathrm{C}_{16} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{O}_{4}$

$$
\begin{aligned}
& V=1533.9(5) \AA^{3} \\
& Z=4 \\
& D_{x}=1.300 \mathrm{Mg} \mathrm{~m}^{-3} \\
& \text { Mo K } \alpha \text { radiation }^{\prime} \\
& \mu=0.10 \mathrm{~mm}^{-1} \\
& T=298(2) \mathrm{K} \\
& \text { Block, yellow } \\
& 0.45 \times 0.40 \times 0.28 \mathrm{~mm}
\end{aligned}
$$

$\gamma=105.017$ (2) ${ }^{\circ}$

## Data collection

Siemens SMART CCD area-
detector diffractometer
$\varphi$ and $\omega$ scans
Absorption correction: multi-scan
(SADABS; Sheldrick, 1996)
$T_{\text {min }}=0.959, T_{\text {max }}=0.974$

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.051$
$w R\left(F^{2}\right)=0.147$
$S=0.96$
5318 reflections
397 parameters

H -atom parameters constrained
$w=1 /\left[\sigma^{2}\left(F_{\mathrm{o}}^{2}\right)+(0.0605 P)^{2}\right]$
where $P=\left(F_{\mathrm{o}}{ }^{2}+2 F_{\mathrm{c}}^{2}\right) / 3$
$(\Delta / \sigma)_{\text {max }}=0.001$
$\Delta \rho_{\text {max }}=0.19 \mathrm{e} \mathrm{A}^{-3}$
$\Delta \rho_{\min }=-0.19 \mathrm{e} \AA^{-3}$

Table 1
Hydrogen-bond geometry ( $\AA \AA^{\circ}$ ).

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :--- | :--- | :--- | :--- | :--- |
| $\mathrm{C} 2-\mathrm{H} 2 B \cdots \mathrm{O}^{\mathrm{i}}$ | 0.97 | 2.57 | $3.392(4)$ | 142 |
| $\mathrm{C} 18-\mathrm{H} 18 B \cdots \mathrm{O} 1$ | 0.97 | 2.46 | $3.394(4)$ | 161 |
| C23-H23 $\cdots 2^{\mathrm{i}}$ | 0.93 | 2.42 | $3.335(4)$ | 168 |
| C26-H26 $_{\mathrm{i}} \mathrm{O}^{\mathrm{ii}}$ | 0.93 | 2.47 | $3.213(5)$ | 137 |

Symmetry codes: (i) $-x,-y,-z+1$; (ii) $-x+1,-y+1,-z+1$.

All H atoms were placed in idealized positions and constrained to ride on their parent atoms, with $\mathrm{C}-\mathrm{H}$ distances in the range $0.93-$ $0.97 \AA$, and with $U_{\text {iso }}(\mathrm{H})$ set to 1.2 or 1.5 times $U_{\text {eq }}(\mathrm{C})$.


Figure 1
The asymmetric unit of (I), showing the atom-labelling scheme. Displacement ellipsoids are drawn at the $30 \%$ probability level.

Data collection: SMART (Siemens, 1996); cell refinement: SAINT (Siemens, 1996); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997a); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997a); molecular graphics: SHELXTL (Sheldrick, 1997b); software used to prepare material for publication: SHELXTL.

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